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Na⁺ uptake and extrusion in the cyanobacterium *Synechocystis* PCC6714 in response to hypersaline treatment. Evidence for transient changes in plasmalemma Na⁺ permeability

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The kinetics of uptake and loss of Na⁺ have been studied using radioisotopic tracer techniques in cells of the cyanobacterium *Synechocystis* PCC6714 exposed to hyperosmotic stress. Cells transferred from a fresh-water-based medium to NaCl at 300–1000 mmol · dm⁻³ showed net Na⁺ uptake during the first 2 min following transfer, with the intracellular Na⁺ level at 2 min increasing as a direct function of the external NaCl concentration. Further incubation of cells in low-level hypersaline media (350–500 mmol · dm⁻³ NaCl) led to a marked reduction in cell Na⁺ within 20 min, indicating an efficient active Na⁺ extrusion system. In contrast, cells maintained in more extreme hypersaline media showed little (750 mmol · dm⁻³ NaCl) or no (1000 mmol · dm⁻³ NaCl) net Na⁺ extrusion following upshock. Cells grown in a saline medium (with NaCl at 500 mmol · dm⁻³) showed a greatly reduced net Na⁺ uptake after 2 min in media containing higher levels of NaCl. However, net Na⁺ uptake was also observed when these cells were downshocked to media containing 50–200 mmol · dm⁻³ NaCl. This is the first demonstration of downshock-induced Na⁺ accumulation in a microbial cell. Time-courses for Na⁺ extrusion in cells downshocked from 500 mmol · dm⁻³ to 100 mmol · dm⁻³ NaCl were similar to those for cells upshocked from freshwater to 500 mmol · dm⁻³ NaCl, requiring approximately 30 min to reach their lowest values. Net Na⁺ extrusion in upshocked cells was found to be markedly sensitive to the external K⁺ concentration, with limited net Na⁺ extrusion in the absence of external K⁺ and maximal reductions in cell Na⁺ in media containing K⁺ at 1–10 mmol · dm⁻³. Temperature was also shown to affect uptake and loss of cell Na⁺ during upshock: cells maintained at 5°C showed no capacity for net Na⁺ extrusion, while higher temperatures (up to 35°C) led to a progressive reduction in the amount of cell Na⁺ at 2 min following upshock and also in the rate of net Na⁺ extrusion after this time.

Introduction

Many cyanobacteria can adjust to hypersaline treatment (upshock) by increasing the intracellular level of a species-specific low molecular weight organic solute. The carbohydrates glucosylglycerol [1], sucrose [2] and trehalose [3], together with the quaternary ammonium compound glycine betaine [4] have been identified as major osmotica in salt-stressed cyanobacteria from freshwater, marine and

hypersaline habitats, with a single organic osmoticum forming the majority of the total low molecular weight organic solute fraction in most cases [5].

Intracellular ion levels have also been measured in several cyanobacteria grown in low-salinity and high-salinity media. In a range of isolates from freshwater, marine and hypersaline environments, including *Synechocystis* PCC6803 [6], *Nodularia harveyana* [7], *Aphanothece halophytica*, *Coc-*

cochloris elabens (*Synechococcus* spp.) and *Dactylococcopsis salina* [4], the internal levels of the monovalent cations K^+ and Na^+ vary only marginally in response to the salinity of the growth medium. In general, the trend is towards higher intracellular K^+ and lower intracellular Na^+ concentrations than in the growth medium, especially when cells are grown in media of elevated salinity [4,8]. Thus the euryhaline unicell *Synechocystis* PCC6803 contains K^+ at approximately $150 \text{ mmol} \cdot \text{dm}^{-3}$ (expressed on a total cell volume basis) while intracellular levels of Na^+ are approximately $35 \text{ mmol} \cdot \text{dm}^{-3}$ when grown in either freshwater or seawater-based media [6]. Such studies, using cells that have been pre-equilibrated for at least 24–48 h in hyperosmotic and hypoosmotic media prior to experimental assay, have given no indication of any involvement of intracellular ions in the long-term osmotic adjustment processes of these cyanobacteria. However, other studies have shown that K^+ may be involved in osmoacclimation in the halotolerant cyanobacterium *Aphanothece halophytica* (*Synechococcus* sp.) with high intracellular K^+ levels (up to $1 \text{ mol} \cdot \text{dm}^{-3}$), varying as a function of the external salt concentration [9]. Further indications that changes in internal ion levels may occur in osmotically-stressed cyanobacteria have also been obtained for the sucrose-accumulating freshwater unicell *Synechococcus* PCC6311 [10]. In this organism, intracellular Na^+ levels rose rapidly following upshock from a freshwater-based medium to a medium containing NaCl at either $300 \text{ mmol} \cdot \text{dm}^{-3}$ or $600 \text{ mmol} \cdot \text{dm}^{-3}$, appearing to reach a level close to that of the external medium within 30 min. Intracellular Na^+ then declined more slowly over the following 16 h, reaching a lower equilibrium level within approx. 40 h. These studies used flame emission spectrophotometric techniques to measure the internal Na^+ levels of pelleted cells, with a correction for external Na^+ present in the adhering medium obtained by using [^{14}C]inulin. This procedure may overestimate intracellular Na^+ since no account is taken of the cation exchange properties of the cell wall matrix [11]. In addition, substances of high molecular weight, including [^{14}C]inulin, may not fully penetrate the cell wall to the plasmalemma, leading to further inaccuracies in the measurement of intracellular solute levels [12].

The additional problems of measuring low internal Na^+ levels against a background of high extracellular Na^+ without adequate rinsing procedures is a further potential source of error [4,9].

We have investigated the effects of upshock upon Na^+ transport and accumulation in the unicellular cyanobacterium *Synechocystis* PCC6714. This euryhaline freshwater isolate grows well in a wide range of salinities, up to and exceeding that of full-strength seawater (full-strength seawater is approx. $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl) and previous studies have shown that glucosylglycerol is involved in the maintenance of osmotic equilibrium and turgor generation in cells grown in seawater-based media [5]. However, changes in intracellular K^+ levels have also been observed in recent experiments, with cells showing a rapid net K^+ uptake in response to hyperosmotic treatment (addition of either NaCl, sorbitol or sucrose at concentrations up to $200 \text{ mmol} \cdot \text{dm}^{-3}$), indicating that a turgor-sensitive K^+ uptake system operates in this cyanobacterium [13], as in some heterotrophic bacteria [14]. In the present study, changes in intracellular ion levels have been monitored over a range of ionic conditions (varying external Na^+ or K^+ concentration) and temperatures, using radioactive tracer techniques to provide further detailed information on the effect of salinity change upon ion uptake and accumulation. The present results show an initial net Na^+ influx (within 2 min) in cells exposed to hyperosmotic and hypoosmotic stress beyond a critical value, followed by K^+ -dependent, temperature-sensitive Na^+ extrusion, leading to a subsequent reduction in cell Na^+ .

Materials and Methods

Cyanobacterium and growth conditions. An axenic isolate of the cyanobacterium *Synechocystis* PCC6714 was obtained from the culture collection of the Institut Pasteur, Paris, France and was maintained in a modified BG11 medium [15] at pH 8.2, with a reduced Na^+ content (as NaNO_3) of $6 \text{ mmol} \cdot \text{dm}^{-3}$ and a K^+ content (as K_2HPO_4) of $0.5 \text{ mmol} \cdot \text{dm}^{-3}$. This medium will be referred to as BG11_m. Cells were grown in 12 dm³ flasks containing 10 dm³ of BG11_m medium at 20°C under continuous illumination (at a photon fluence rate of $45 \mu\text{mol} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$). Cells were harvested

in later exponential phase, at biovolume densities approaching $1 \text{ mm}^3 \cdot \text{cm}^{-3}$.

Upshock experiments were carried out by adding NaCl, to give a range of hyperosmotic media from $50 \text{ mmol} \cdot \text{dm}^{-3}$ to $1000 \text{ mmol} \cdot \text{dm}^{-3}$. The effects of varying external K^+ concentration were studied by the addition of KCl to the medium, to give external K^+ concentrations up to $10 \text{ mmol} \cdot \text{dm}^{-3}$. Temperature control was achieved using thermostatically-operated water baths; in experiments where the cyanobacterium was subjected to changes in ambient temperature, cells were pre-incubated for 30 min at each temperature prior to upshock ($500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl).

Measurement of intracellular volumes. Cell volumes of *Synechocystis* PCC6714 were determined using a particle size analyzer (Coulter Electronics Ltd., Luton, U.K., model ZB (industrial), fitted with a C1000 channelyzer unit) linked to an Acorn microcomputer to provide direct estimation of cell density, mean cell volume and biovolume density. A $50 \mu\text{m}$ aperture (path length $60 \mu\text{m}$) was used throughout.

In experiments where the cells were upshocked, membrane-filtered medium of identical composition and osmotic strength was used in the particle size analyzer, to ensure no change in cell volume during measurement. Care was taken to replace the entire contents of the aperture tube and associated glassware with the experimental medium, to ensure osmotic and ionic homogeneity during counting. Cells in basal medium (BG11_m) and in low-NaCl media (up to $100 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl) were counted and sized in the minimum amount of electrolyte required for adequate functioning of the Coulter counter system, i.e. $9 \text{ g} \cdot \text{dm}^{-3}$ NaCl. For each value of mean cell volume, six replicate samples were taken, with 10 000–18 000 cells per sample volume (50 mm^3).

Radioisotopic measurement of Na^+ exchange. Na^+ uptake and loss were studied using the radioisotope $^{24}\text{Na}^+$ (obtained from the Scottish Universities Research and Reactor Centre, E. Kilbride, U.K.). All samples were counted in 5-cm^3 aliquots of scintillation cocktail 229 (Packard Instruments, IL, U.S.A.) using a Packard Instruments 300 liquid scintillation spectrometer, with automatic quench correction (samples channels ratio). All data were corrected for decay of $^{24}\text{Na}^+$ during counting since

$^{24}\text{Na}^+$ has a half-life of 15 h.

Preliminary time course experiments were carried out using cells maintained in BG11_m medium containing $^{24}\text{Na}^+$ (at specific activities up to $5 \text{ GBq} \cdot \text{mol}^{-1}$) for 18 h to allow exchange and equilibration of intracellular and extracellular Na^+ [4]. Cells were subsequently upshocked by the addition of radioisotopically-labelled NaCl containing $^{24}\text{Na}^+$ at the same specific activity and then incubated in experimental media for varying times, after which the cells were separated from the medium by a rapid filtration procedure since silicone oil microcentrifugation techniques proved unsatisfactory due to problems associated with changes in medium density upon upshock [16]. Aliquots ($0.5\text{--}2.0 \text{ cm}^3$) of cell suspensions were removed from the experimental flasks and transferred to 'nuflow' cellulose acetate membrane filters (Oxoid Ltd., Basingstoke, U.K.) of pore size $0.45 \mu\text{m}$ and diameter 25 mm. Adhering medium was then removed from the cells by vacuum filtration. Cells were then rinsed rapidly with identical medium, free of added $^{24}\text{Na}^+$, to remove any extracellular radioisotope present in the remaining medium and associated with cation exchange groups on the cell wall. Preliminary experiments using cells that had been killed by pre-treatment in 50% dimethyl sulphoxide for 30 min showed that over 98% of the extracellular $^{24}\text{Na}^+$ fraction was removed by rinsing with $12\text{--}15 \text{ cm}^3$ of non-radioactive medium over a period of 25–30 s. All subsequent experiments were carried out using the same filtration and rinsing procedures, although pre-equilibration in $^{24}\text{Na}^+$ -containing BG11_m medium was not required in all cases due to the low intracellular Na^+ content of such cells (see below).

Experiments were carried out in a Chandos Instruments (New Mills, U.K.) multi-place stirrer base, at 20°C under constant illumination (at a photon fluence rate of $45 \mu\text{mol} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$) unless otherwise stated.

Results

Effects of hyperosmotic stress upon Na^+ uptake and accumulation

Fig. 1 shows a time-course for changes in cell Na^+ upon transfer of *Synechocystis* PCC6714 from BG11_m medium containing $^{24}\text{Na}^+$ (18 h pre-in-

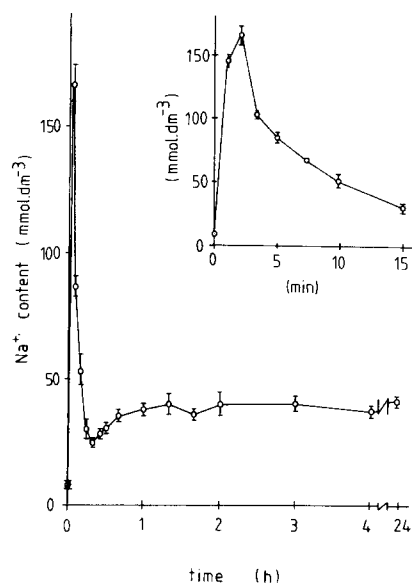


Fig. 1. Intracellular Na^+ content of *Synechocystis* PCC6714 transferred from freshwater BG11_m to a medium containing $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl. Values (measured using $^{24}\text{Na}^+$ equilibration technique) represent the mean of three replicates (\pm S.D.); insert shows short-term changes (up to 15 min).

cubation) to a hyperosmotic medium containing NaCl (plus $^{24}\text{Na}^+$ at the same specific activity) at $500 \text{ mmol} \cdot \text{dm}^{-3}$. The initial value for cell Na^+ in BG11_m medium was low, being less than $10 \text{ mmol} \cdot \text{dm}^{-3}$. However, upon upshock, cell Na^+ was observed to rise rapidly to a maximum value of $167 \text{ mmol} \cdot \text{dm}^{-3}$ within 2 min. A less rapid decline in cell Na^+ was then observed over the next 18 min, reaching a value close to the original Na^+ concentration at 20 min. Subsequent incubation led to a further small increase in cell Na^+ , with values up to $40 \text{ mmol} \cdot \text{dm}^{-3}$ being recorded between 2 h and 24 h (Fig. 1). These values are significantly higher than the value at 20 min (t -test $P = 0.05$) suggesting that the increase in Na^+ content observed after 20 min was not due solely to experimental variation. Furthermore, the same pattern was also observed in all subsequent experiments at this concentration (see below).

Fig. 2a contains data for net Na^+ uptake in cells of *Synechocystis* PCC6714 transferred from BG11_m medium to medium containing $^{24}\text{Na}^+$ -labelled NaCl at 50–1000 $\text{mmol} \cdot \text{dm}^{-3}$ for 2 min (see Fig. 1). Minimal uptake of external Na^+ was

observed below $200 \text{ mmol} \cdot \text{dm}^{-3}$. However, a direct relationship between Na^+ uptake and external Na^+ concentration was observed in more saline media (i.e. between $300 \text{ mmol} \cdot \text{dm}^{-3}$ and $1000 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl), with the highest values for internal Na^+ (approx. $680 \text{ mmol} \cdot \text{dm}^{-3}$, expressed on a total cell volume basis) being recorded in $1000 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl. Over this range, the cells showed minimal change in volume in response to hypersaline treatment (Fig. 2a).

Cells of *Synechocystis* PCC6714 grown in a saline medium (BG11_m plus additional NaCl at $500 \text{ mmol} \cdot \text{dm}^{-3}$) for a period of 10 days showed limited Na^+ uptake in hyperosmotic media (i.e. $> 500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl), with a 3-fold reduction in intracellular $^{24}\text{Na}^+$ level upon transfer to $1000 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl (Fig. 2b). However, these cells also showed net Na^+ uptake when exposed to hyposmotic shock greater than $-200 \text{ mmol} \cdot$

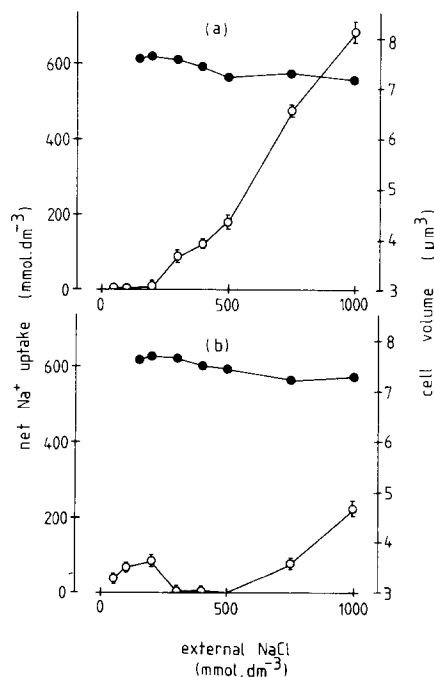


Fig. 2. Net uptake of Na^+ (as $^{24}\text{Na}^+$) and cell volume of *Synechocystis* PCC6714 grown in (a) freshwater BG11_m and (b) BG11_m plus NaCl at $500 \text{ mmol} \cdot \text{dm}^{-3}$ after 2 min in a range of media from 500 to $1000 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl. Values for cell Na^+ uptake (\circ) represent the mean of three replicates (\pm S.D.); cell volume data (\bullet) were obtained using six replicate measurements (S.D. $< 0.2 \mu\text{m}^3$ in all cases).

dm^{-3} , with significant $^{24}\text{Na}^+$ accumulation after a 2 min period in media containing NaCl at 50–200 $\text{mmol} \cdot \text{dm}^{-3}$. Cell volume showed similar, small changes to freshwater-grown cells in response to upshock and downshock (Fig. 2b).

Uptake and loss of $^{24}\text{Na}^+$ were also studied in media containing NaCl at 350–1000 $\text{mmol} \cdot \text{dm}^{-3}$ (Fig. 3). Cells in 350 $\text{mmol} \cdot \text{dm}^{-3}$ and 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl (Fig. 3a) showed a similar pattern of Na^+ influx and efflux following upshock, although the levels of cell Na^+ were substantially lower in 350 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl than in 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl. Further increases in external Na^+ all resulted in higher values for intracellular Na^+ at 2 min with diminishing reductions in cell Na^+ concentration between 2 min and 60 min being observed at higher salinities. Thus cells transferred to 750 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl showed limited net Na^+ loss during this period while cells in 1000 $\text{mmol} \cdot \text{dm}^{-3}$ were found to increase their intracellular Na^+ concentration up to 60 min (Fig. 3b). Cell volume changes were minimal during the time-

courses shown in Fig. 3 (data not shown). It appears that the ability of cells to actively lower their intracellular Na^+ content following upshock is governed by the extent of hyperosmotic treatment.

Uptake and loss of Na^+ in cells exposed to hypoosmotic shock

Fig. 2 showed that cells of *Synechocystis* PCC6714 grown in BG11_m plus additional NaCl at 500 $\text{mmol} \cdot \text{dm}^{-3}$ increased their internal level of Na^+ upon transfer to hypoosmotic media (i.e. with NaCl at 50–200 $\text{mmol} \cdot \text{dm}^{-3}$). A time-course for this response is shown in Fig. 4 for cells transferred to 100 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl. A transient increase in internal Na^+ was observed within the first 2 min following downshock, reaching a maximum value of 64 $\text{mmol} \cdot \text{dm}^{-3}$ (Fig. 4). Subsequent incubation in 100 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl led to a reduction in intracellular Na^+ , reaching a new steady-state value in approx. 30 min. The response of cells grown in saline media to downshock into 100 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl is thus similar in many respects to the behaviour of freshwater-grown *Synechocystis* PCC6714 upon upshock into 350–500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl (cf. Figs. 3 and 4).

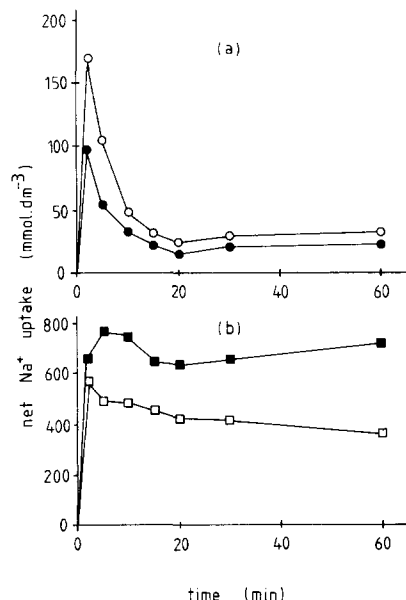


Fig. 3. Net Na^+ uptake in cells of *Synechocystis* PCC6714 transferred to a medium containing NaCl at 350–1000 $\text{mmol} \cdot \text{dm}^{-3}$. Values (three replicates, S.D. < 15 $\text{mmol} \cdot \text{dm}^{-3}$) are shown in (a) for 350 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl (●) and 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl (○); (b) contains data for 750 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl (□) and 1000 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl (■).

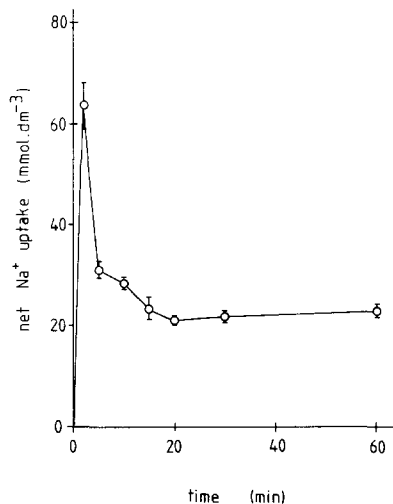


Fig. 4. Net Na^+ uptake in downshocked cells of *Synechocystis* PCC6714 transferred from BG11_m plus NaCl at 500 $\text{mmol} \cdot \text{dm}^{-3}$ to a medium containing 100 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl for up to 60 min. Values represent the mean of three replicates (\pm S.D.).

Effects of varying external K^+ concentration upon Na^+ uptake and loss in response to hyperosmotic treatment

Preliminary experiments demonstrated that the response shown in Fig. 1 for cells in $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl was insensitive to changes in external pH, with similar increases and decreases in cell Na^+ being observed at pH 6 and pH 9 (data not shown). However, the responses of *Synechocystis* PCC6714 were found to be sensitive to variations in external K^+ concentration (Fig. 5). Thus, when cells were upshocked in $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl in the absence of K^+ in their bathing medium, the initial rise in internal Na^+ up to 2 min was followed by a limited decrease in cell Na^+ up to 60 min. In contrast, cells that were upshocked in $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl medium containing K^+ at 1.0 or $10.0 \text{ mmol} \cdot \text{dm}^{-3}$ showed similar major reductions in cell Na^+ between 2 and 20 min (Fig. 5). Cells transferred to $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl plus $0.5 \text{ mmol} \cdot \text{dm}^{-3}$ K^+ (the K^+ concentration in BG11_m medium, and in all other experiments, see

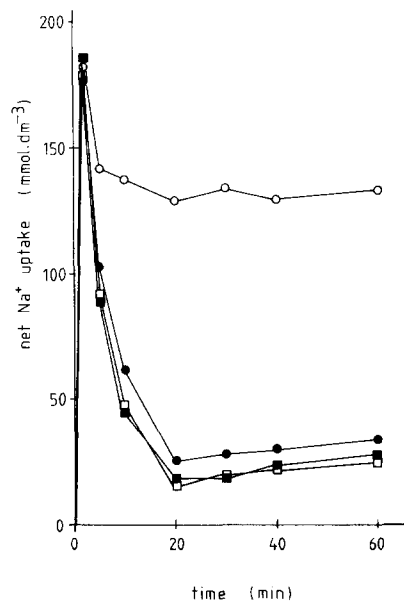


Fig. 5. Effects of varying external K^+ concentration upon Na^+ uptake and extrusion in *Synechocystis* PCC6714 transferred from fresh water BG11_m to a medium containing NaCl at $500 \text{ mmol} \cdot \text{dm}^{-3}$. Values (three replicates; S.D. $< 8 \text{ mmol} \cdot \text{dm}^{-3}$) are shown for external K^+ levels of $0.0 \text{ mmol} \cdot \text{dm}^{-3}$ (○), $0.5 \text{ mmol} \cdot \text{dm}^{-3}$ (●), $1.0 \text{ mmol} \cdot \text{dm}^{-3}$ (□) and $10.0 \text{ mmol} \cdot \text{dm}^{-3}$ (■).

Figs. 1–4) also showed substantial net Na^+ extrusion within 20 min, although the values at this K^+ concentration were found to be somewhat higher than in media containing 1.0 and $10.0 \text{ mmol} \cdot \text{dm}^{-3}$ K^+ . Thus external K^+ concentrations in excess of $0.5 \text{ mmol} \cdot \text{dm}^{-3}$ were necessary for maximum net Na^+ extrusion following hyperosmotic treatment ($500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl).

Effects of temperature upon Na^+ uptake and loss in response to hyperosmotic treatment

Fig. 6 shows data for the internal Na^+ content of *Synechocystis* PCC6714 transferred from BG11_m medium to a medium containing $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl at temperatures between 5°C and 35°C . At the lowest temperature (5°C) a steady increase in intracellular Na^+ uptake was observed following the initial rapid phase of Na^+ uptake (2 min), with maximum values for intracellular Na^+ being recorded after a 60 min incubation period at 5°C . Cells at 15 – 35°C all showed the same biphasic response, with increasing cell Na^+ up to 2 min and

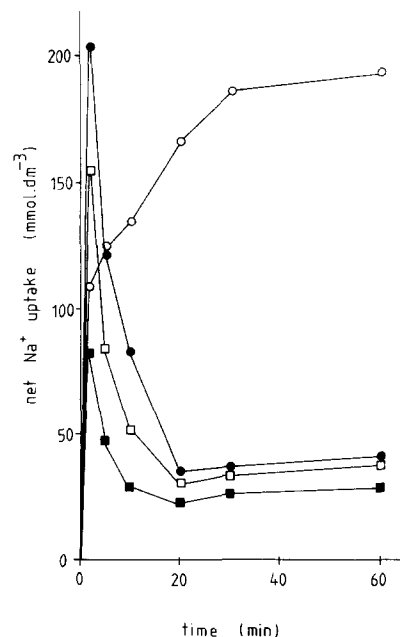


Fig. 6. Effects of temperature upon Na^+ uptake and extrusion in *Synechocystis* PCC6714 transferred from fresh water BG11_m to a medium containing NaCl at $500 \text{ mmol} \cdot \text{dm}^{-3}$. Values (three replicates; S.D. $< 9 \text{ mmol} \cdot \text{dm}^{-3}$) are shown for cells incubated at 5°C (○), 15°C (●), 25°C (□) and 35°C (■).

decreasing cell Na^+ between 2 and 20 min (with smaller, subsequent increases up to 60 min). However, maximal changes in intracellular Na^+ were observed at 15–20°C while minimal increases and decreases were recorded for cells incubated at 35°C. The values for intracellular Na^+ in cells transferred to 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl at 35°C (Fig. 6) are thus lower than those for cells upshocked in 350 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl at 20°C (Fig. 3).

Discussion

The data in Fig. 1 show that intracellular Na^+ (as measured by $^{24}\text{Na}^+$ equilibration for 18 h) is low in freshwater-grown *Synechocystis* PCC6714 ($< 10 \text{ mmol} \cdot \text{dm}^{-3}$ Na^+ prior to upshock). This value was not exceeded in the long-term, even after 48 h incubation in radioactively-labelled BG11_m medium (data not show). Thus steady-state equilibration was reached within 18 h. Low intracellular Na^+ concentrations are characteristic of non-vacuolate, eukaryotic algal cells [17,18] and *Synechocystis* PCC6714 contains Na^+ at levels comparable to the lowest values reported for algal cells. The low intracellular Na^+ content of *Synechocystis* PCC6714 is also consistent with current theories regarding solute compartmentation in photosynthetic cells, with vacuolar localization of (toxic) Na^+ and cytoplasmic accumulation of K^+ together with organic solutes [19] since cyanobacterial cells lack large vacuoles.

The data in Fig. 2a show that cells of *Synechocystis* PCC6714 reduced their volume by less than 10% upon transfer from a freshwater-based medium to medium containing NaCl at up to 1000 $\text{mmol} \cdot \text{dm}^{-3}$. Such a response is in direct conflict with the assumption that cells transferred to hyperosmotic media behave as so-called 'ideal osmometers', shrinking to an extent that is directly proportional to the reciprocal of the external osmolality once a critical (minimum) osmolality is exceeded and cell turgor has been removed [20,21]. This study suggests that NaCl would not be a suitable solute to use in studies of the changes in cell volume that accompany increases in osmolality, due to net solute uptake at high external NaCl concentration (Fig. 2). Thus, changes in cell volume due to water flux from the cells in response to

upshock were mitigated by solute transfer across the plasmalemma, since a substantial net Na^+ uptake was observed in hyperosmotic media (300–1000 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl). Below 300 $\text{mmol} \cdot \text{dm}^{-3}$ there was little evidence of substantial Na^+ penetration of the cell interior and the cells would thus be expected to behave in the manner predicted by 'classical' osmotic theory [20] within this range. This may also explain why such phenomena have not been observed previously, using low- Na^+ media [22,23]. Above 300 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl, the direct relationship between net Na^+ uptake and external NaCl concentration will have served to minimize any decrease in cellular volume due to upshock (Fig. 2a), with cells transferred to 1000 $\text{mmol} \cdot \text{dm}^{-3}$ showing the greatest proportional increase in net Na^+ uptake.

Acclimation of *Synechocystis* PCC6714 in a saline medium is also demonstrated by the data contained in Fig. 2, since cells grown in 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl accumulated less $^{24}\text{Na}^+$ than freshwater-grown cells upon transfer to extreme hypersaline media (i.e. $> 500 \text{ mmol} \cdot \text{dm}^{-3}$). Such effects may help to explain how the upper salinity tolerance limit of an organism can be increased by pre-treatment at an intermediate salinity [24]. Previous studies have concentrated on the changes in low molecular weight organic solutes that occur in cyanobacteria grown in saline media, with the inference that these compounds may act as 'compatible solutes' [25], protecting the cell from NaCl inactivation of enzyme activity and loss of metabolic function [26]. However, the present study suggests that acclimation may also be related to direct ionic effects and to changes in intracellular ion levels upon upshock.

The observation that cells grown in a saline medium (500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl) showed net accumulation of Na^+ when downshocked to low-salinity media is a novel demonstration of the effects of hypoosmotic stress (Figs. 2 and 4). Previous studies have concentrated on intracellular ion increases during upshock and this phenomenon does not appear to have been reported previously. Net Na^+ uptake was induced by hypoosmotic shocks in excess of $-200 \text{ mmol} \cdot \text{dm}^{-3}$. This value compares favourably with the NaCl change required to initiate Na^+ uptake during hyperosmotic treatment (Fig. 2), suggesting that it is the extent

of osmotic (NaCl) shock, rather than its direction, which is responsible for the change in plasmalemma permeability, resulting in net Na^+ accumulation. The present study also suggests that transient disruption of cellular metabolism, photosynthetic activity, etc. during downshock may also be related to the penetration of Na^+ into the cell interior. Previous work using micro-celled algae has stressed osmotic rather than ionic effects during downshock [27].

Time-course data for cells upshocked to 350 $\text{mmol} \cdot \text{dm}^{-3}$ and 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl show rapid entry of Na^+ within 2 min, indicating a large increase in membrane permeability to Na^+ during this period, followed by slower net efflux, leading to a lower post-upshock value at 20 min (Fig. 3a). These data can be used to estimate net Na^+ efflux rates over this time period of $28.2 \text{ nmol} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ (expressed relative to plasmalemma surface area) [13] in 350 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl and $50.4 \text{ nmol} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ in 500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl. On a cell volume basis, the rates are comparable to the net Na^+ effluxes reported for Na^+ -enriched cells of *Chlorella pyrenoidosa* upon transfer to K^+ -enriched medium [28,29] at $5\text{--}8 \text{ mmol} \cdot \text{dm}^{-3} \cdot \text{min}^{-1}$. However, the net efflux of Na^+ , complete within 20 min in 350–500 $\text{mmol} \cdot \text{dm}^{-3}$ NaCl is considerably faster than that reported by Blumwald et al. [10] for the cyanobacterium *Synechococcus* PCC6311, where intracellular Na^+ remained high for up to 16 h. Rapid net Na^+ extrusion following upshock-induced increases in internal Na^+ has also been observed in the euryhaline eukaryotic alga *Dunaliella tertiolecta* [30] with new steady-state values being established within 30 min. Downshocked cells of *Synechocystis* PCC6714 also extrude Na^+ within a similar time period (at a rate of approx. $14.6 \text{ nmol} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$; Fig. 4).

Cells of *Synechocystis* PCC6714 showed an upper limit for net Na^+ extrusion of approx. $750\text{--}1000 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl (Fig. 3b). Since *Synechocystis* PCC6714 is able to grow in seawater-based media and in hypersaline media up to this level of NaCl, there may be a direct link between growth and the capacity to actively extrude Na^+ upon upshock.

The observation that net extrusion of Na^+ is sensitive to external K^+ , with little evidence of

sustained Na^+ efflux in media with no added K^+ (Fig. 5) suggests that some form of exchange between Na^+ and K^+ is involved. Batterton et al. [31] have provided evidence that a $(\text{Na}^+/\text{K}^+)\text{-ATPase}$ may exist in cyanobacterial cells, although the activities quoted in their study of *Anacystis nidulans* (*Synechococcus* sp.) suggest that such a component would form a relatively minor fraction of the total ATPase activity. The insensitivity of net Na^+ extrusion to changes in external pH also suggests that the response does not rely upon a Na^+/H^+ antiport system, such as that proposed for *Anacystis nidulans* (*Synechococcus* sp.) [17, 23, 32] and *Synechococcus* PCC6311 [33].

The effects of added external K^+ upon net Na^+ extrusion in *Synechocystis* PCC6714 are comparable to the reports of net Na^+ efflux upon addition of K^+ in Na^+ -enriched *C. pyrenoidosa* [28] where K^+ -depleted cells accumulated Na^+ to high intracellular levels (in excess of $75 \text{ mmol} \cdot \text{dm}^{-3}$). Addition of K^+ at $3 \text{ mmol} \cdot \text{dm}^{-3}$ led to a rapid decrease in cell Na^+ , reaching less than $10 \text{ mmol} \cdot \text{dm}^{-3}$ within 90 min. This time-course is comparable to the data shown in Fig. 1 for *Synechocystis* PCC6714 in $500 \text{ mmol} \cdot \text{dm}^{-3}$ NaCl. Similarly, cells of the unicellular alga *Dunaliella tertiolecta* show transient upshock-induced Na^+ uptake, followed by rapid, K^+ -sensitive net Na^+ extrusion [30].

Previous studies have shown that K^+ influx in *Synechocystis* PCC6714 is greatly stimulated (by a factor of up to 3-fold) upon addition of NaCl at $200 \text{ mmol} \cdot \text{dm}^{-3}$ to the bathing medium, with a consequent increase in intracellular K^+ content within 20 min [13]. These changes indicate that a turgor-sensitive K^+ transport system operates in this cyanobacterium since, at concentrations up to $200 \text{ mmol} \cdot \text{dm}^{-3}$, net Na^+ accumulation will be minimal (Fig. 2a). The possibility of a stoichiometric exchange of (intracellular) Na^+ and (extracellular) K^+ in more extreme hypersaline media remains to be tested and will be further complicated by the existence of turgor-sensitive K^+ uptake.

The effects of temperature upon Na^+ uptake and loss in hyperosmotic media have not hitherto been reported, with decreasing Na^+ accumulation as a function of increasing temperature (Fig. 6). The response is not related to loss of viability at high temperature, since *Synechocystis* PCC6714

has been grown routinely at 35°C in this laboratory and 37°C is the optimum temperature for growth of many cyanobacteria. Recent studies have also shown that carbohydrate accumulation in this strain is temperature-sensitive, with cells grown at high temperature showing considerable accumulation of a second carbohydrate, sucrose, in addition to glucosylglycerol [34]. The possibility that the effects of temperature upon ion and organic solute levels in *Synechocystis* PCC6714, are linked requires further investigation.

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